Listing of Claims

This listing of claims supersedes all previous listings of claims.

What is Claimed:

- 1. (Canceled)
- (Previously presented) The fuel cell according to claim 62, wherein the hydrocarbon is a petroleum distillate.
- (Previously presented) The fuel cell according to claim 2, wherein the petroleum distillate is selected from the group consisting of gasoline, diesel oil, naphtha, JP-4, JP-5, JP-8, kerosene, motor oil, natural gas, fuel oil, and mixtures thereof.
- (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of JP-4, JP-5, JP-8, and mixtures thereof.
- (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of naptha, kerosene, fuel oil, and mixtures thereof.
- (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of gasoline, diesel oil, natural gas, and mixtures thereof.
- (Original) The fuel cell according to claim 2, wherein the hydrocarbon comprises an alcohol.
- (Currently amended) The fuel cell according to claim 7, wherein the alcohol is selected from the group consisting of methanol, ethanol, and mixtures thereof comprises ethanol.
- 9. (Previously presented) The fuel cell according to claim 2, wherein the hydrocarbon is

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selected from the group consisting of butane, toluene, decane, and mixtures thereof.

- 10. (Previously presented) The fuel cell according to claim 62, wherein the sulfur containing hydrocarbon fuel has a sulfur content of from about 1 ppm to about 1000 ppm.
- 11. (Previously presented) The fuel cell according to claim 10, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 10 ppm to about 1000 ppm.
- 12. (Previously presented) The fuel cell according to claim 11, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 20 ppm to about 1000 ppm.
- 13. (Previously presented) The fuel cell according to claim 12, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 100 ppm to about 1000 ppm.
- 14. (Previously presented) The fuel cell according to claim 13, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 250 ppm to about 1000 ppm.
- 15. (Previously presented) The fuel cell system according to claim 62 wherein the solid electrolyte is an oxide ion conducting material.
- 16. (Previously presented) The fuel cell according to claim 15, wherein the oxide ion conducting material is selected from the group consisting of doped ceria, doped zirconia, and doped lanthanum gallate.
- 17. (Previously presented) The fuel cell according to claim 16, wherein the doped ceria is selected from the group consisting of gadolinium doped ceria, samarium-doped ceria, yttria-doped ceria, and mixtures thereof.
- 18. (Previously presented) The fuel cell according to claim 15, wherein the oxide ion conducting material is yttria-doped zirconia.

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- (Previously presented) The fuel cell according to claim 16, wherein the doped zirconia is scandium-doped zirconia.
- 20. (Canceled)
- 21 (Previously presented) The process according to claim 63, wherein the hydrocarbon is a petroleum distillate.
- 22. (Previously presented) The process according to claim 21, wherein the petroleum distillate is selected from the group consisting of gasoline, diesel oil, naphtha, JP-4, JP-5, JP-8, kerosene, motor oil, natural gas, fuel oil, and mixtures thereof.
- 23. (Previously presented) The process according to claim 22, wherein the petroleum distillate is selected from the group consisting of JP-4, JP-5, JP-8, and mixtures thereof.
- 24. (Previously presented) The process according to claim 22, wherein the petroleum distillate is selected from the group consisting of naphtha, kerosene, fuel oil, and mixtures thereof
- 25. (Original) The process according to claim 22, wherein the petroleum distillate comprises gasoline.
- 26. (Original) The process according to claim 22, wherein the petroleum distillate comprises diesel oil.
- 27. (Previously presented) The process according to claim 63, wherein the hydrocarbon is selected from the group consisting of alcohols, butane, toluene, decane, and mixtures thereof.
- 28. (Original) The process according to claim 27, wherein the hydrocarbon comprises an Page 4 of 11

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alcohol

- (Currently amended) The process according to claim 28, wherein the alcohol is selected
 from the group consisting of methanol, ethanol, and mixtures thereof comprises ethanol.
- 30. (Previously presented) The process according to claim 63, wherein the sulfur containing hydrocarbon has a sulfur content of from about 10 ppm to about 1000 ppm.
- 31. 53. (Canceled without prejudice)
- 54. (Canceled)
- 55. (Previously presented) The fuel cell system of claim 62, wherein the anode further comprises copper deposited in the pores.
- 56. (Previously presented) The process of claim 63, wherein the anode further comprises copper deposited in the pores.
- 57. (Canceled)
- 58. (Canceled) The fuel cell system of claim 64, wherein the anode further comprises copper deposited in the pores.
- 59. (Canceled)
- 60. (Canceled) The process of claim 66, wherein the anode further comprises copper deposited in the pores.
- 61. (Canceled)

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62. (Previously presented) A solid oxide fuel cell system capable of directly operating with a sulfur-containing hydrocarbon fuel that does not undergo prior treatment to remove organic sulfur compounds, comprising:

(a) a solid electrolyte comprising an electronic insulator that allows transfer of anions:

(b) an essentially nickel-free porous anode containing at least ceria deposited in the pores, the anode further comprising a ceramic, and at least a portion of the anode being bound to the electrolyte:

(c) a cathode:

(d) a fuel comprising a hydrocarbon having 2 or more carbons, and the fuel being characterized as having a sulfur content of from about 1 ppm to about 5000 ppm; and

(e) an oxygen source;

wherein the solid electrolyte and the porous anode overlap one another so as to define a region of physical contact between one another, the region of physical contact being characterized as an essentially uninterrupted interface.

63. (Previously presented) A process of producing electrical energy, comprising:

(a) providing a solid oxide fuel cell system capable of directly operating with a sulfurcontaining hydrocarbon fuel that does not undergo prior treatment to remove organic sulfur compounds, the solid oxide fuel cell system comprising:

a solid oxide electrolyte comprising an electronic insulator that allows transfer of anions:

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> an essentially nickel-free porous anode containing at least ceria deposited in the pores, the anode further comprising a ceramic, and at least a portion of the anode being bound to the electrolyte; and

a cathode, and

a fuel comprising a hydrocarbon having two or more carbons, and the fuel being characterized as having a sulfur content of from about 1 ppm to about 5000 ppm,

wherein the solid electrolyte and the anode overlap one another so as to define a region of physical contact between one another, the region of physical contact being characterized as an essentially uninterrupted interface.

- (b) contacting the cathode with an oxygen source; and
- (c) contacting the porous anode with the fuel.
- 64. (Previously presented) A solid oxide fuel cell system capable of directly operating with a sulfur-containing hydrocarbon fuel that does not undergo prior treatment to remove organic sulfur compounds, comprising:
- (a) a solid electrolyte comprising an electronic insulator that allows transfer of anions:
- (b) an essentially nickel-free porous anode containing at least copper deposited in the pores, the anode further comprising a ceramic, and at least a portion of the anode being bound to the electrolyte;
- (c) a cathode;

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(d) a fuel comprising a hydrocarbon having 2 or more carbons, and the fuel being characterized as having a sulfur content of from about 1 ppm to about 5000 ppm; and

(e) an oxygen source;

wherein the solid electrolyte and the porous anode overlap one another so as to define a region of physical contact between one another, the region of physical contact being characterized as an essentially uninterrupted interface.

65. (Canceled)

66. (Canceled) A process of producing electrical energy, comprising:

(a) providing a solid oxide fuel cell system capable of directly operating with a sulfurcontaining hydrocarbon fuel, the solid oxide fuel cell comprising

a solid oxide electrolyte that is an electronic insulator that allows transfer of anions,

an essentially nickel-free porous anode, the anode further comprising a ceramic, the anode containing at least <u>copper</u> eeria deposited in the pores and comprising a porous ceramic, and at least a portion of the anode being bound to the electrolyte, and

a cathode.

(b) contacting said cathode with an oxygen source; and

(c) contacting said porous anode with a fuel comprising a hydrocarbon having two or more carbons, the fuel being characterized as having a sulfur content of from about 1 ppm to about 5000 ppm,

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wherein the solid electrolyte and the porous anode overlap one another so as to define a region of physical contact between one another, the region of physical contact being characterized as an essentially uninterrupted interface.

- 67. (Canceled)
- 68. (Canceled) The process according to claim 66, wherein the hydrocarbon is selected from the group consisting of alcohols, a petroleum distillate, butane, toluene, decane, or any combination thereof.